Class 1 : Electric Forces

- Electric charge
- Computing electric forces: Coulomb's Law, superposition
- What is the electric field?
- Properties of field lines

• Electric charge is an intrinsic property of the particles which make up matter, which experimentally can be either *positive or negative*





 Electric charge is quantized such that protons and electrons have equal and opposite charge e = ± 1.6 × 10⁻¹⁹ C [unit C = Coulombs]



Electric charge is locally conserved and cannot be created or destroyed



• Two electric charges attract or repel each other with equal and opposite forces



• The strength of the force is given by **Coulomb's Law**



- The force is proportional to the magnitude of the charges q_1 , q_2
- The force is inversely proportional to the square of the separation *r*
- The force acts along the line joining the charges : by symmetry, no other direction could be singled out
- The force strength is governed by the **permittivity of free space** ε_0 where $\frac{1}{4\pi\varepsilon_0} = 9 \times 10^9 Nm^2C^{-2}$

Vectors

- A vector is a quantity such as a force which has both a magnitude and a direction
- It can be indicated by \vec{F} or \underline{F} or F (helpful!)
- A vector can be specified by its **components** along coordinate axes, such as $\vec{F} = (F_x, F_y, F_z)$
- The magnitude of a vector is $\left|\vec{F}\right| = \sqrt{F_x^2 + F_y^2 + F_z^2}$
- A **unit vector**, indicated by \vec{F} , has magnitude = 1
- You may also see $\vec{F} = F_x \hat{\iota} + F_y \hat{j} + F_z \hat{k}$, in terms of unit vectors along co-ordinate axes

Vectors

- Vectors may be **added** by summing their components
- The **dot product** of two vectors, $\vec{a} \cdot \vec{b}$, is the projection of one vector along the other, $|\vec{a}||\vec{b}|\cos\theta$ ($\theta = angle between$). It can also be evaluated as $\vec{a} \cdot \vec{b} = a_x b_x + a_y b_y + a_z b_z$
- The **cross product** of two vectors, $\vec{a} \times \vec{b}$, is a vector perpendicular to both with magnitude $|\vec{a}||\vec{b}| \sin \theta$



How strong are electromagnetic forces?

$$F_{g} = G \frac{m_{1}m_{2}}{r^{2}} \qquad F_{e} = k_{e} \frac{q_{1}q_{2}}{r^{2}}$$

$$G = 7 \times 10^{-11} N kg^{-2} m^{2} \qquad k_{e} = 9 \times 10^{9} N C^{-2} m^{2}$$

- What is the ratio between the gravitational force F_g and the electrostatic force F_e between two isolated electrons (mass $9 \times 10^{-31} kg$, charge $1.6 \times 10^{-19} C$)?
- Does your answer depend on the distance between the electrons?

 Electromagnetism is by far the strongest force in our everyday experience



• Electrostatic forces are reduced in practice because matter is **approximately electrically neutral**



• Other forces dominate on different scales ...



Principle of superposition

• The total force from multiple charges is given by the **principle of superposition**



Clicker question

Two charges +q and -q are on the y-axis, symmetric about the origin. In what direction is the force on a positive charge on the x-axis?



A. Up B. Down

C. Left





Principle of superposition

• What is the force on the charge at the origin?



 Coulomb's Law tells us that *like charges repel and* unlike charges attract. But what causes one charge to feel the effect of the other?



- How is a force transmitted across space?
- Is a force transmitted instantaneously and, if so, does that violate special relativity?

- How is a force *communicated* between charges? A useful model is the **electric field**
- We interpret that *electric charges set up an electric* field \vec{E} in the region of space around them
- Then, a "test charge" q, placed in the electric field, will feel a force $\vec{F} = q \vec{E}$
- The electric field is a vector "force field" representing the size/direction of the force per unit charge

Please note in workbook

• The electric field can be represented by **field lines**:



• Electric field lines start on positive charges and end on negative charges. Their *direction* shows the force acting on a test charge; their *spacing* indicates the force strength. *They can never cross.*

• A more complicated example of an electric field:



 The electric field around a point charge +Q follows simply from Coulomb's Law



 Place a test charge +q at distance r from +Q

• Force
$$\vec{F} = \frac{Qq}{4\pi\varepsilon_0 r^2} \vec{\hat{r}}$$

• Electric field
$$ec{E}=rac{ec{F}}{q}=rac{Q}{4\piarepsilon_0r^2}ec{\hat{r}}$$

Try example in workbook

Clicker question

An electron is placed in an electric field. In which direction does it move?



- A. It remains stationary
- B. It moves in the direction of the electric field lines
- C. It moves in the opposite direction to the electric field lines
- D. It moves in another direction



Clicker question



Summary

- Electric charge is a fundamental property of nature
- Electric charges q_1 , q_2 at a distance r attract/repel with a **Coulomb force** $\vec{F} = \frac{q_1 q_2}{4\pi \varepsilon_0 r^2} \vec{\hat{r}}$
- Forces from different charges may be combined using the principle of vector superposition
- Electric charges set up an **electric field** \vec{E} in the region of space around them; a test charge q placed in the field feels a force $\vec{F} = q\vec{E}$